

## OVERCOMING CHALLENGES OF LED IN GRAPHIC ARTS APPLICATIONS

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## **INDUSTRY EVOLUTION FROM UV LAMP TO LED**

Market value of UV LEDs for the curing application continuing to grow<sup>1</sup>

- Estimated 2016~2021 growth US\$81M to US\$195M
- Market penetration forecasted to reach 50~60% by 2020

Technical/Economic/Environmental advantages to LED

#### ✤ ~75% Lower power consumption

Current consumption of the UV systems

~120 kV

~30.5 kW

UV curing process (7 steps incl. 2 grinding stations and 1 final curing) of a parquet flooring:

Gallium iodide and mercury medium pressure lamps (blue) had been compared with a combination of UV LEDs and high-performance UVC lamps (green).

The kW-numbers show the difference in the consumption of electricity which is significantly lower in the "green" system. There is an energy saving of about 75 percent. No warm up, fast start-up

- Appropriate for temperature sensitive substrates
- Environmentally friendlier
  - No ozone production
  - Longer lifetime
  - No Hg concerns (Hg Arc Lamps)

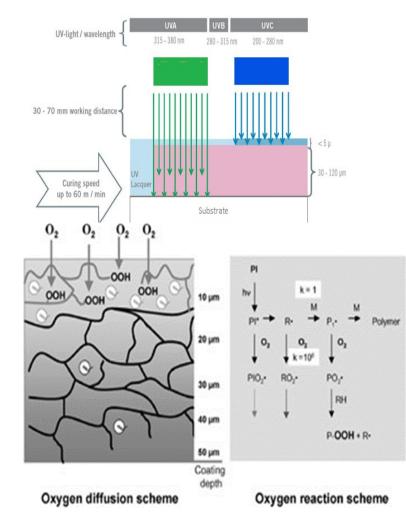


#### LED INDUSTRY CHALLENGE

- →Narrow wavelength emission reduced to 365nm,395nm,405nm
- $\rightarrow$  Reduction in surface curing
  - Due to oxygen inhibition
  - 200-280nm best for surface cure

#### $\rightarrow$ ~5um at surface remains uncured

- Peroxy radicals form when oxygen reacts with free radicals
- Oxygen terminates PI and acrylate radicals
- Exacerbated in thin films, O<sub>2</sub> continually diffuses



UV Coatings: Basics, Recent Developments and New Applications, Chapter 7. Reinholds Schwalm, Elsevier 2007.



## STRATEGIES FOR COUNTERING OXYGEN INHIBITION

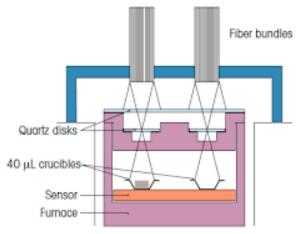
#### → Mechanical solutions

- Nitrogen inert
- Ensure maximum distance of LED is < 2cm to substrate
- $\rightarrow$  Optimize photo-initiator selection and concentration
  - Match PI absorbance with LED transmission
  - Excess PI can also inhibit cure, ladder study to confirm loadings
- $\rightarrow$  Practical guide for chemistry based solutions
  - Increase formulation reactivity
  - Optimize double bond concentration
  - Reduce glass transition effect
  - Hydrogen donors for peroxy radical conversion
    - Thiols > Amines > Ethers
    - Dr. Jon Scholte, "Thiolene Chemistry, past and progress" presentation Tue 5/8/18

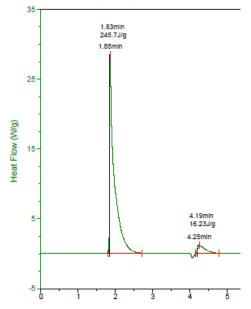


## ANALYTICAL TECHNIQUES: MEASURING REACTIVITY

- $\rightarrow$  Heat is generated during polymerization
- → (△H) Enthalpy, can be quantified using Photo-Differential Scanning Calorimeter (DSC)
- → Sensors below crucibles detect and measure heat changes
  - Empty crucible serves as control
  - LED/UV focused over samples
  - Exposure typically pulsed ~5-10 sec
- $\rightarrow$  Sum of integrated peak(s) give J/g



Cross-section of the optical system and the DSC furnace

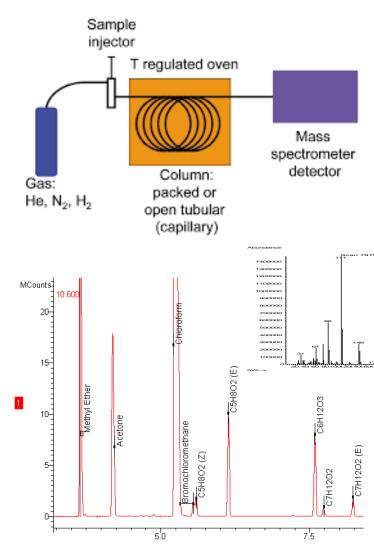


SARTOM



## **ANALYTICAL TECHNIQUES: RESIDUAL MONOMER**

- → Residual, unreacted monomer left post cure indicates degree of conversion
- → Residual monomer quantification by GC-Mass Spectrometry
  - Cured samples immersed in solvent
  - Extract injected, monomers elucidated and quantified using GC/MS
- →Limitations on MW for Gas Chromatography (~550 m/z)
  - Analytes must be volatile





## **REACTIVITY & CONVERSION STUDY**

#### $\rightarrow$ Evaluated low & high molecular weight

polyester & polyurethane acrylates

- 2f LMW PEA
- 6f HMW PEA
- 6f LMW PUA
- 2f HMW PUA
- 16f PEA Hyper Branched PEA
- Increasing molecular weight and functionality monomers
- $\rightarrow$  Effects of ethoxylation in light blue

Monomer	Dendrimer PEA	High MW PEA	Low MW PEA	Low MW UA	Low MW UA
	Wt%	Wt%	Wt%	Wt%	Wt%
PhEA	75/20	75/20	75/20	75/20	75/20
HDDA	75/20	75/20	75/20	75/20	75/20
TCDMDA	75/20	75/20	75/20	75/20	75/20
ТМРТА	75/20	75/20	75/20	75/20	75/20
ΡΕΤΑ	75/20	75/20	75/20	75/20	75/20
DI-PETA	75/20	75/20	75/20	75/20	75/20
(3EO)TMPTA	75/20	75/20	75/20	75/20	75/20
(9EO) TMPTA	75/20	75/20	75/20	75/20	75/20
PI (50/50 TPO/BAPO)	5	5	5	5	5



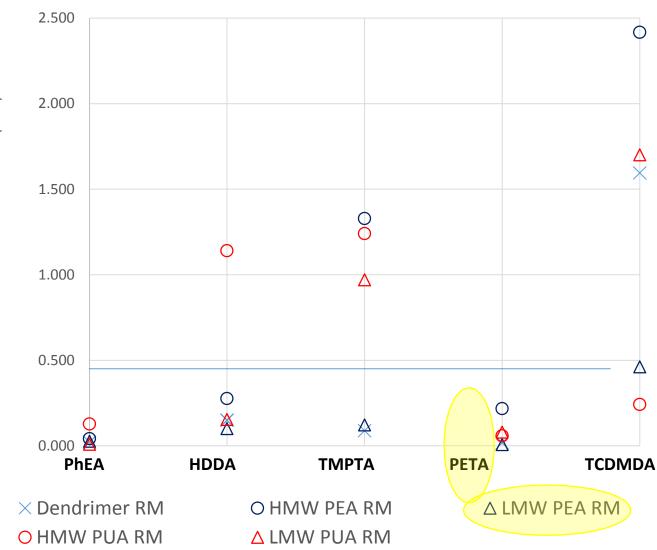
## **INCREASE REACTIVITY THROUGH FUNCTIONALITY**

 $\rightarrow$  (C<sub>db</sub>), Total double bond concentration is important <sup>2</sup>

- PETA > Di-PETA
- $\rightarrow$  (C<sub>db</sub>) mol/L = f<sub>m</sub>(d<sub>m</sub>/MW<sub>m</sub>)
  - f = # of acrylate functionality
  - d<sub>m</sub>= density(g/mL)
  - MW<sub>m</sub> = molecular weight (g/mol)
- → Residual monomer results confirm PETA had highest conversion
  - Lowest TMPTA and TCDMDA
  - Additional properties at work!

f	1	2	2	3	4	5
Monomer	PhEA	TCDMDA	HDDA	тмрта	PETA	DI-PETA
	[C <sub>db</sub> ] mol/L					
NEAT						
Monomer	5.8	7.2	8.9	11	13	11
Dendimer PEA	6.0	7.2	8.5	9.9	12	10
Low MW PEA	5.4	6.5	7.9	9.3	12	9.3
High MW PEA	5.2	6.3	7.7	9.1	11	9.1
Low MW UA	6.2	7.3	8.7	10	12	10
High MW UA	5.4	6.5	7.9	9.2	11	9.3

#### **RESIDUAL MONOMER: CLOSER LOOK**



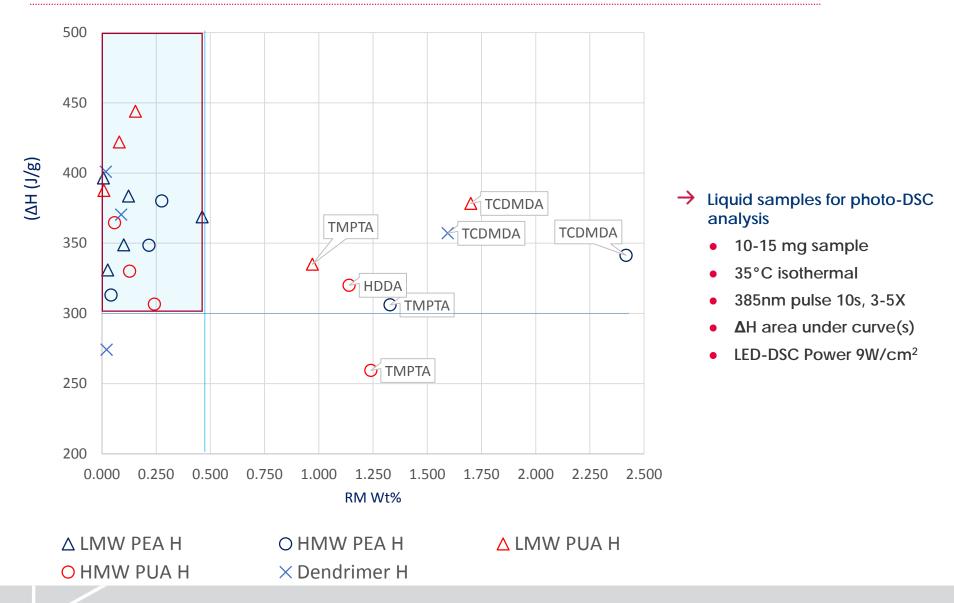
#### **RM** sample preparation

- #6 Wire rod draw down on Al
  - Cure speed 50 fpm X 2
- 395nm LED
- LED Power 12W/cm<sup>2</sup>
- 0.2g sample to 5mL acetone
   24hr extraction
- 10 ppm Quantification Limit
- Varian Ion Trap GC/MS





#### RESIDUAL MONOMER VERSUS HEAT OF REACTION (J/g)





## **OBSERVATIONS ON INCREASING REACTIVITY**

 $\rightarrow$  Majority of residual monomer results <0.3% (Wt.)

- Analysis did not include acrylated or inert impurities
- → Majority of heat of reaction for formulations with low residual monomer >300 J/g
  - LMW PEA diluted in PETA lowest at 10 ppm
- → High residual monomers outliers included formulations with TCDMDA, TMPTA
  - Also had >300 J/g
- → Caveat emptor!
  - Great screening tool
  - Need additional data to better differentiate products for LED

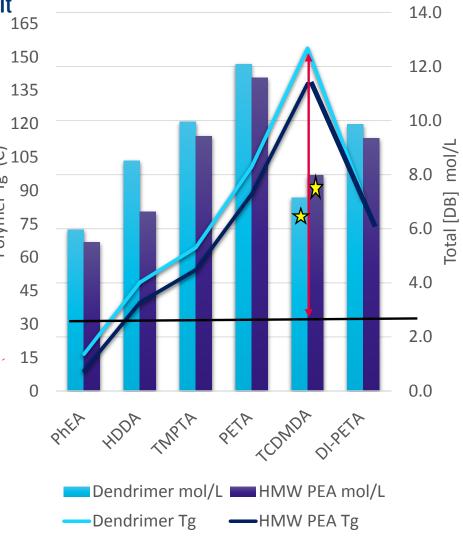


## **INVESTIGATING EFFECT OF TG: POLYESTER ACRYLATE**

# → Products with high glass transitions limit nolecule mobility Faster path to vitrification Faster path to vitrification TCDMDA combination of high Tg 105 and low C<sub>db</sub> Good indicator of slower cure

 $\rightarrow$  PETA C<sub>db</sub> compensates for Tg effect

- Verified by residual monomer
- → PhEA exhibited great conversion
  - Tg below (close-to)ambient cure temperature
  - >0.04 %(Wt) residual monomer





## INVESTIGATING EFFECT OF TG: POLYURETHANE ACRYLATE

#### $\rightarrow$ Exaggerated Tg effect for TCDMDA

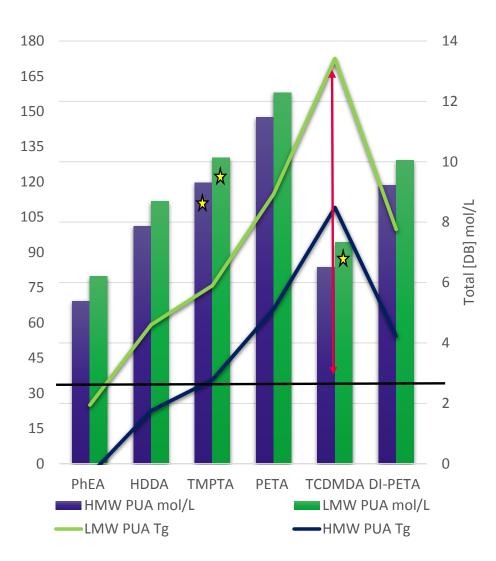
 High Tg, (6f) LMW PUA combination dramatically slowed reaction (< 270 J/g)

#### 

- Expected poor performance from high Tg LMW PUA
- Potential issue with high initial viscosity HMW PUA

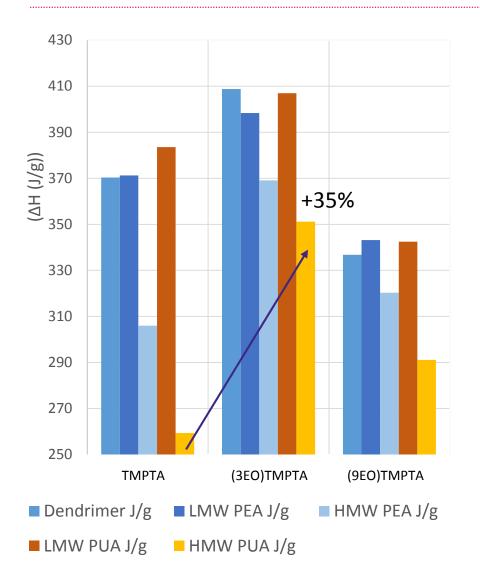
#### $\rightarrow$ PhEA

>0.13%(Wt) residual monomer





## HEAT OF REACTION (J/g) & ETHOXYLATION



Monomer	Dendrimer PEA	High MW PEA	Low MW PEA	Low MW UA	Low MW UA
	Wt%	Wt%	Wt%	Wt%	Wt%
(3EO)TMPTA	75/20	75/20	75/20	75/20	75/20
(9EO) TMPTA	75/20	75/20	75/20	75/20	75/20
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 $\rightarrow$  (3EO)TMPTA increases reactivity

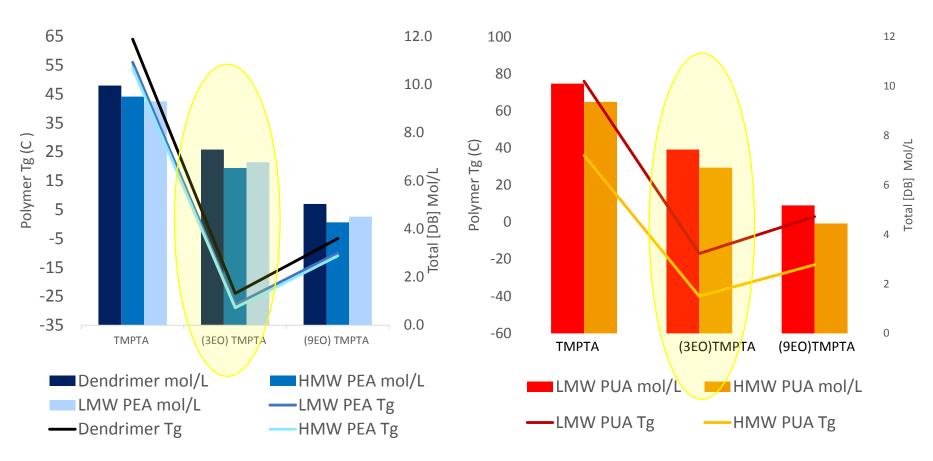
- Increases seen in all formulations
- ~35% increase for the HMW PUA
- → (9EO)TMPTA no significant improvement
  - More EO, not necessarily better, diminished double bond concentration



## EFFECT OF T<sub>G</sub> VERSUS [C<sub>DB</sub>] & ETHOXYLATION

#### → Trend of highest reactivity remains consistent

- Optimize double bond concentration
- Reduced polymer glass transition



#### FLEXO INK FORMULATIONS: ACRYLATED AMINES/POLYTHIOLS

→ Industry recognized "Thumb Twist" to evaluate surface cure quality

• Convenient



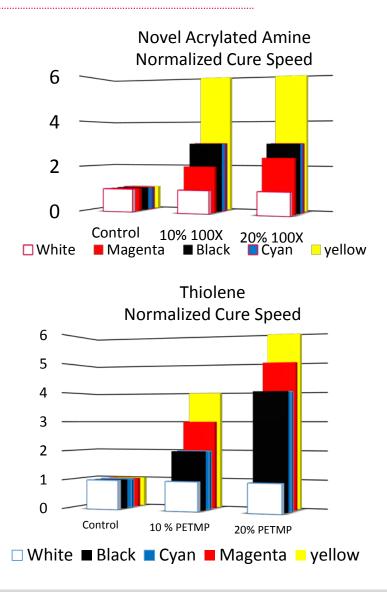
• Subjective

Amine modified PEA increased cure speed (fpm)

 10-20% loadings cured 2-6x's the belt speed

→ PETMP cure speed at 20% PETMP loading

- 2 6 x's cure speed, color dependent
- Stability issues at high loadings



Base Formualtion: Equal Parts of 4f Polyester, PO NPGDA & Di Penta Tetra Acrylate, ~15% pigment loadings **SARTOMER** LED Peak Irradience 12W/cm2,395nm Note: White cured at max line speed (500 fpm)

#### CONCLUSIONS

#### $\rightarrow$ Photo-DSC convenient instrument for evaluating heat of reaction

- Need additional data to better differentiate products for LED
- Much more power analytical tool when in parallel residual monomer

#### $\rightarrow$ Practical ways to boost formulation reactivity

- Optimize double bond concentration
- Balance functionality with molecular weight of the system
- Balance high/low glass transition materials

#### → Optimize base formulation before adding hydrogen donors

 Ethoxylation is least efficient (Thiols > Amines > Ethers), had positive contribution



## THANK YOU FOR YOUR ATTENTION QUESTIONS?



